# SDS

# SAFETY DATA SHEET

# Oakwood Products, Inc 730 Columbia HWY N Estill, SC 29918 www.oakwoodchemical.com

Phone Numbers:

Product Information	803-739-8800
Transportation Emergency	800-451-8346
Outside the USA	760-602-8700

# MATERIAL IDENTIFICATION

NAME: Iron(III) chloride, anhydrous CAS#: [7705-08-0] CAT#: 099039 For R&D use only.

# HAZARDS IDENTIFICATION

# **GHS Classification**

Corrosive to Metals (Category 1) Acute toxicity, oral (Category 4) Skin corrosion/irritation (Category 2) Serious eye damage/eye irritation (Category 1) Hazardous to the aquatic environment, acute hazard (Category 2)

# GHS Label elements, including precautionary statements

Pictograms

Signal Word



Danger

Hazard Statement(s)			
H290	May be corrosive to metals		
H302	Harmful if swallowed		
H315	Causes skin irritation		
H318	Causes serious eye damage		
H401	Toxic to aquatic life		
Precautionary Statement(s)			

P234	Keep only in original container.
P280	Wear protective gloves/protective clothing/eye protection/face protection.
P301 + P330 + P331	IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.
P302 + P352	IF ON SKIN: wash with plenty of soap and water.

P304 + P340	IF INHALED: Remove victim to fresh air and Keep at rest in a position comfortable for breathing.
P305 + P351 + P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
P310	Immediately call a POISON CENTER or doctor/physician.
P332 + P313	IF SKIN irritation occurs: Get medical advice/attention.

# **COMPOSITION/INFORMATION ON INGREDIENTS**

Synonyms	:	Molysite, Iron trichloride, Ferric chloride
Formula	:	FeCl3
Molecular Weight	:	162.21 g/mol

CAS	Description	Concentration
7705-08-0	Iron(III) chloride, anhydrous	98%

#### FIRST AID MEASURES

#### In case of eye contact

Immediately flush eyes with running water for at least 15 minutes while keeping eyes open. Seek medical attention.

# In case of skin contact

Wash thoroughly with soap and plenty of water. Seek medical attention.

#### If inhaled

Remove victim from source of exposure to fresh air. If breathing is difficult, administer oxygen. Seek medical attention.

#### If swallowed

Do not induce vomiting. Give water to victim to drink. Seek medical attention.

# FIRE-FIGHTING MEASURES

#### Suitable extinguishing media

Use carbon dioxide, dry chemical powder, alcohol-resistant or polymer foam.

#### Special protective equipment for fire-fighters

Wear self-contained breathing apparatus and protective clothing to prevent contact with skin and eyes.

#### Unusual fire and explosion hazards/decomposition of product

emits toxic fumes under fire conditions.

# ACCIDENTAL RELEASE MEASURES

#### **Personal precautions**

Use personal protective equipment. Avoid breathing fumes, vapors, mists or gas. Ventilate area. Remove all sources of ignition. Evacuate personnel.

#### **Environmental precautions**

Prevent further leakage if safe to do so.

#### Methods and materials for containment and clean up

Absorb spills on sand or vermiculite and place in closed container for disposal.

# HANDLING AND STORAGE

## Precautions for safe handling

Avoid prolonged use. Avoid all direct contact with material. Do not breathe dust or vapor. Wash thoroughly after handling. Reacts violently with water.

#### Precautions for safe storage

Keep container tightly closed. Store in a cool, dry, well-ventilated area.

# EXPOSURE CONTROL/PERSONAL PROTECTION

Contains no substances with occupational exposure limit values.

# Personal protective equipment

#### Eye/face protection

Wear protective safety goggles or face shields tested and approved under appropriate government standards such as NIOSH (US) or EN 166 (EU).

#### Hand/skin protection

Avoid all direct contact with product. Wear chemical-resistant gloves. Wear protective clothing and boots. After contact with skin, wash immediately.

#### **Respiratory protection**

Ensure adequate ventilation during use. Approved respiratory equipment must be used when airborne concentrations are unknown or exceed the exposure limits.

# PHYSICAL AND CHEMICAL PROPERTIES

Appearance	Dark grey powder
Odour	no data available
Odour Threshold	no data available
Melting point/Freezing Point	304°C
Boiling Point	no data available
Flash Point	no data available
Evaporation Rate	no data available
Flammability (solid, gas)	no data available
Upper/Lower Flammability or Explosive limits	no data available
Vapour pressure	no data available
Relative Density	no data available
Solubility(ies)	no data available
Partition coefficient: n-octanol/water	no data available
Auto-ignition temperature	no data available
Decomposition temperature	no data available

Viscosity Refractive Index

## STABILITY AND REACTIVITY

#### **Chemical stability**

Stable under recommended storage conditions.

# Possibility of hazardous reactions

Exothermic in contact with water Forms shock-sensitive mixtures with certain other materials

# Conditions to avoid

no data available

# **Incompatible materials** Strong oxidizing agents, strong bases, and potassium, alkali metals.

#### Hazardous decomposition products

May evolve carbon monoxide, carbon dioxide, hydrogen chloride, and hydrogen fluoride.

# TOXICOLOGICAL INFORMATION

Acute toxicity

Oral LD50 Mouse - 1,300 mg/kg

Dermal LD50 Rabbit - > 2,000 mg/kg

Skin corrosion/irritation Rabbit - Irritating

Serious eye damage/eye irritation Rabbit - Severe irritating

**Respiratory or skin sensitization** no data available

Germ cell mutagenicity no data available

#### Carcinogenicity no data available

Reproductive toxicity no data available

# **STOT-single exposure** no data available

# **STOT-repeated exposure** no data available

Iron(III) chloride, anhydrous

# Aspiration hazard

no data available

## **Exposure Routes**

Causes burns. May have harmful effects if inhaled or swallowed.

# Signs and Symptoms of Exposure

Spasm, inflammation and edema of the larynx, spasm, inflammation and edema of the bronchi, pneumonitis, pulmonary edema. Overdose of iron compounds may have a corrosive effect on the gastrointestinal mucosa and be followed by necrosis, perforation, and stricture formation. Several hours may elapse before symptoms that can include epigastric pain, diarrhea, vomiting, nausea, and hematemesis occur. After apparent recovery a person may experience metabolic acidosis, convulsions, and coma hours or days later. Further complications may develop leading to acute liver necrosis that can result in death due to hepatic coma.

# **Additional Information**

RTECS: LJ9100000

To the best of our knowledge, the health hazards of this material have not been fully investigated.

# **ECOLOGICAL INFORMATION**

#### Toxicity

Toxicity to fish: LC50 - Pimephales promelas (fathead minnow) - 21.84 mg/l - 96 h

Toxicity to daphnia and other aquatic invertebrates: EC50 - Daphnia magna (Water flea) - 9.6 mg/l - 48 h

# Persistence and degradability

no data available

#### **Bioaccumulative potential**

no data available

#### Mobility in soil no data available

# PBT and vPvB assessment

no data available

## Other adverse effects

no data available

#### DISPOSAL CONSIDERATIONS

Dissolve in or mix the material with a combustible solvent and burn in a chemical incinerator equipped with an afterburner and scrubber. Observe all Federal, State and local laws.

#### TRANSPORT INFORMATION

**DOT** Ferric chloride, anhydrous 8 UN1773 III Reportable Quantity (RQ): 1000 lbs

# IMDG

Ferric chloride, anhydrous 8 UN1773 III EMS-No: F-A, S-B Marine Pollutant: Yes

# IATA

Ferric chloride, anhydrous 8 UN1773 III

# **REGULATORY INFORMATION**

## SARA 302 Components

SARA 302: No chemicals in this material are subject to the reporting requirements of SARA Title III, Section 302.

# SARA 313 Components

SARA 313: No chemicals in this material are subject to the reporting requirements of SARA Title III, Section 313.

#### New Jersey Right to Know Components

This product contains a chemical on the New Jersey Right to Know Components List.

Iron(III) chloride, anhydrous

# California Prop. 65 Components

This product may contain a chemical known to the State of California to cause cancer, birth defects, or other reproductive harm.

CAS

7705-08-0

# OTHER INFORMATION

Version: 1.0

Revision Date : 2/5/2016

The above information is believed to be correct but does not purport to be all-inclusive and shall be used only as a guide. Oakwood shall not be held liable for any damage resulting from handling or from contact with the above product.